

INTEGRATION OF BIOTECHNOLOGY AND CATALYSIS



FOR THE VALORIZATION OF WASTE STREAMS

Tomorrow's
biorefineries
in Europe

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Partners and expertise

TUHH

Hamburg University of Technology

An-Ping Zeng

Biotechnologies

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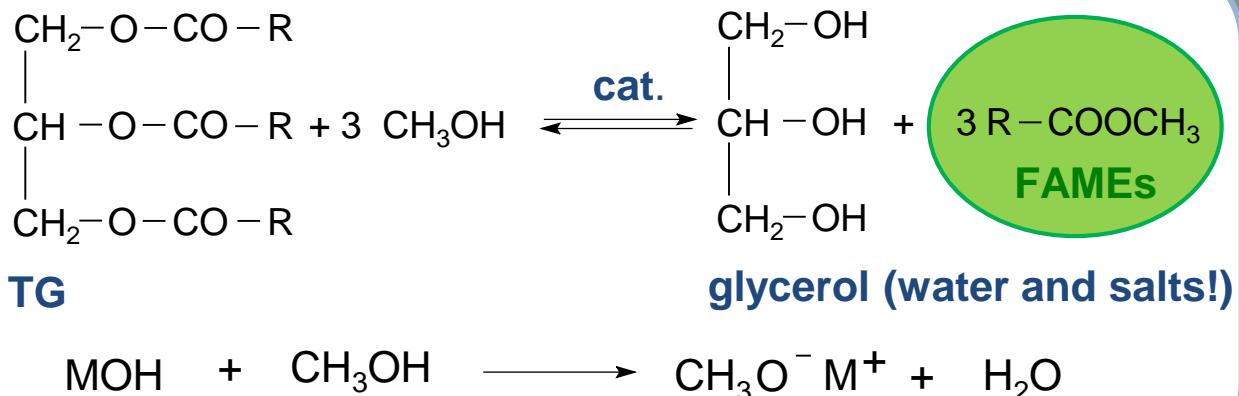
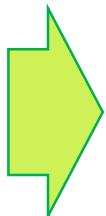
Consorzio
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Michele Aresta, Angela Dibenedetto
Chemical technologies

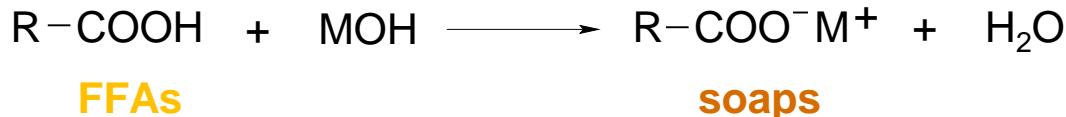
Luigi Palmieri, Gennaro Agrimi
Biotechnologies

ARKEMA

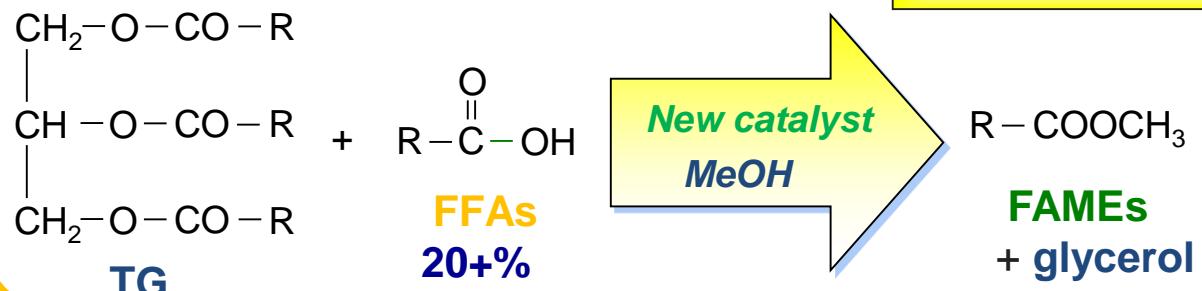




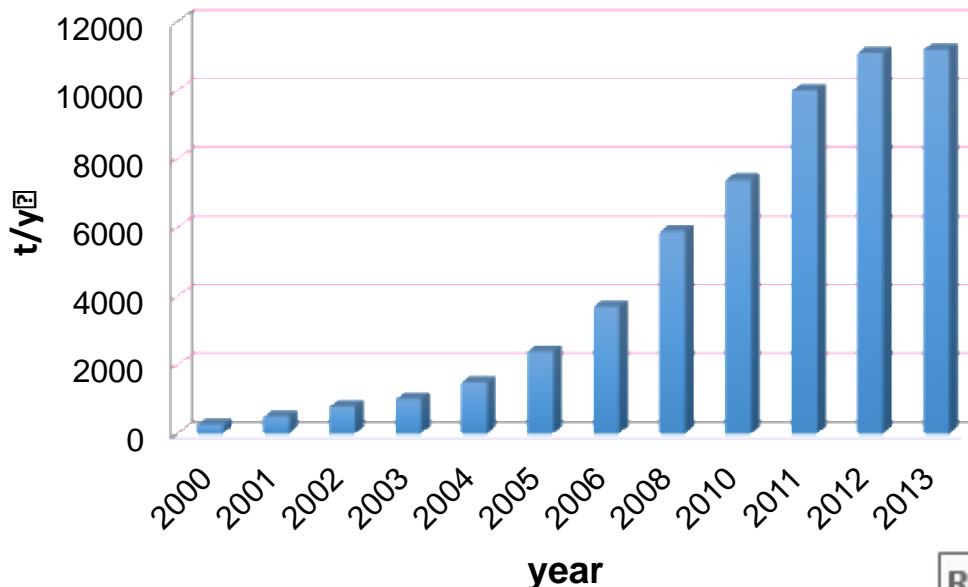
Watery



Water free



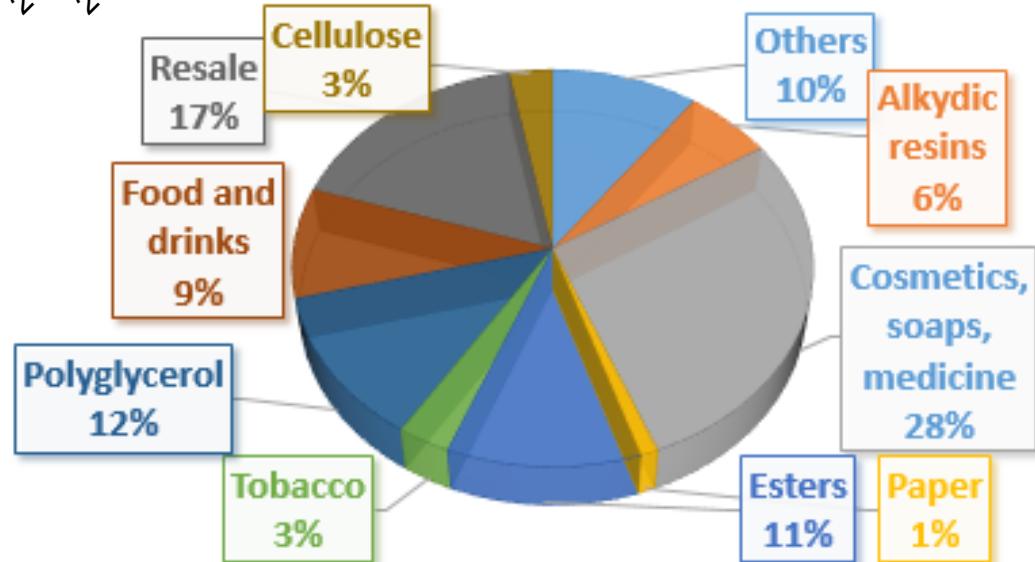
M. Aresta, A. Dibenedetto,
A. Angelini, C. Pastore,
L. di Bitonto, "New catalysts
for the production of
biodiesel from bio-oils",
Patent MI2013A001730, 2013



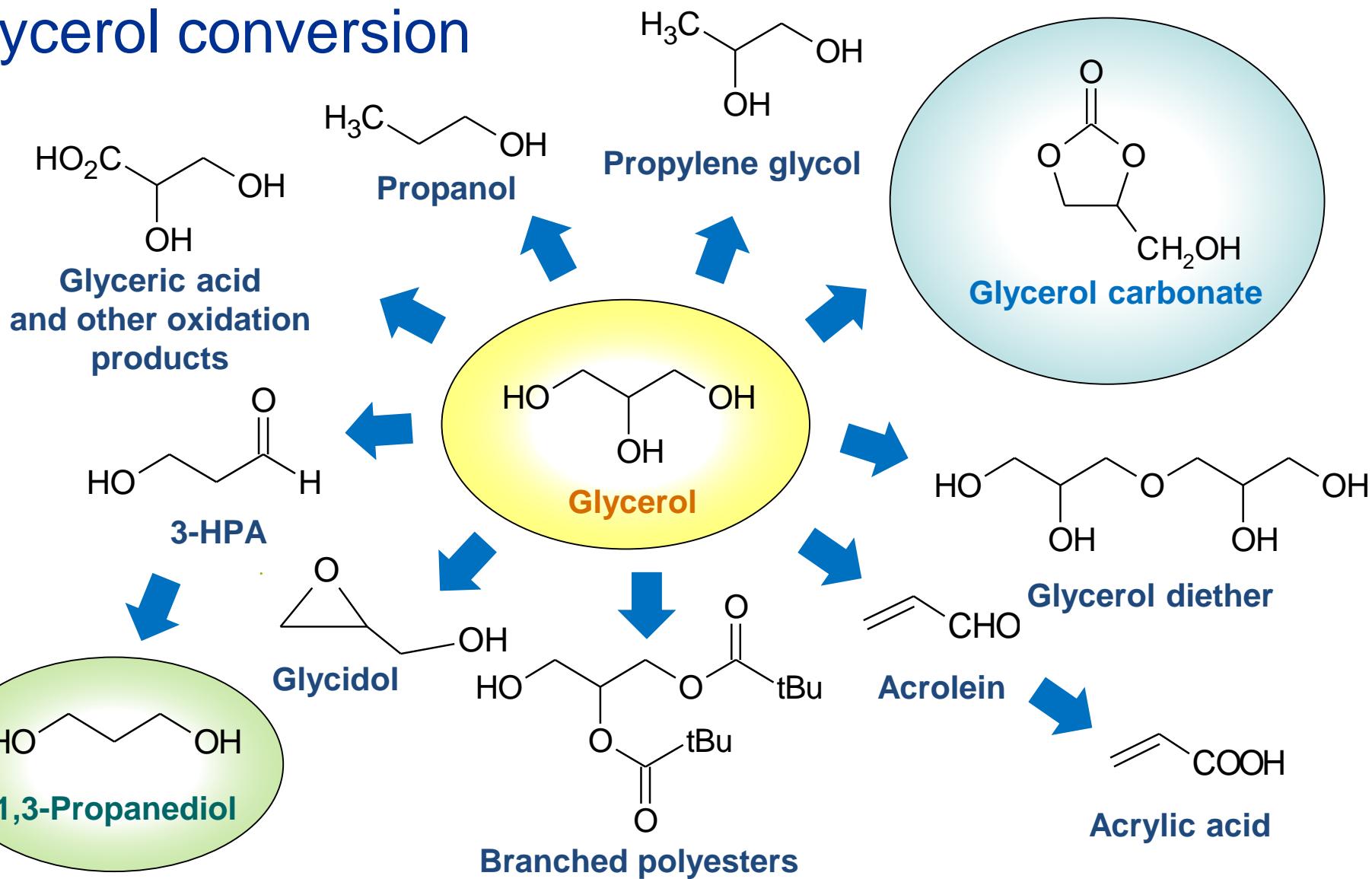
EU market of bioglycerol and uses

EU production of bioglycerol.

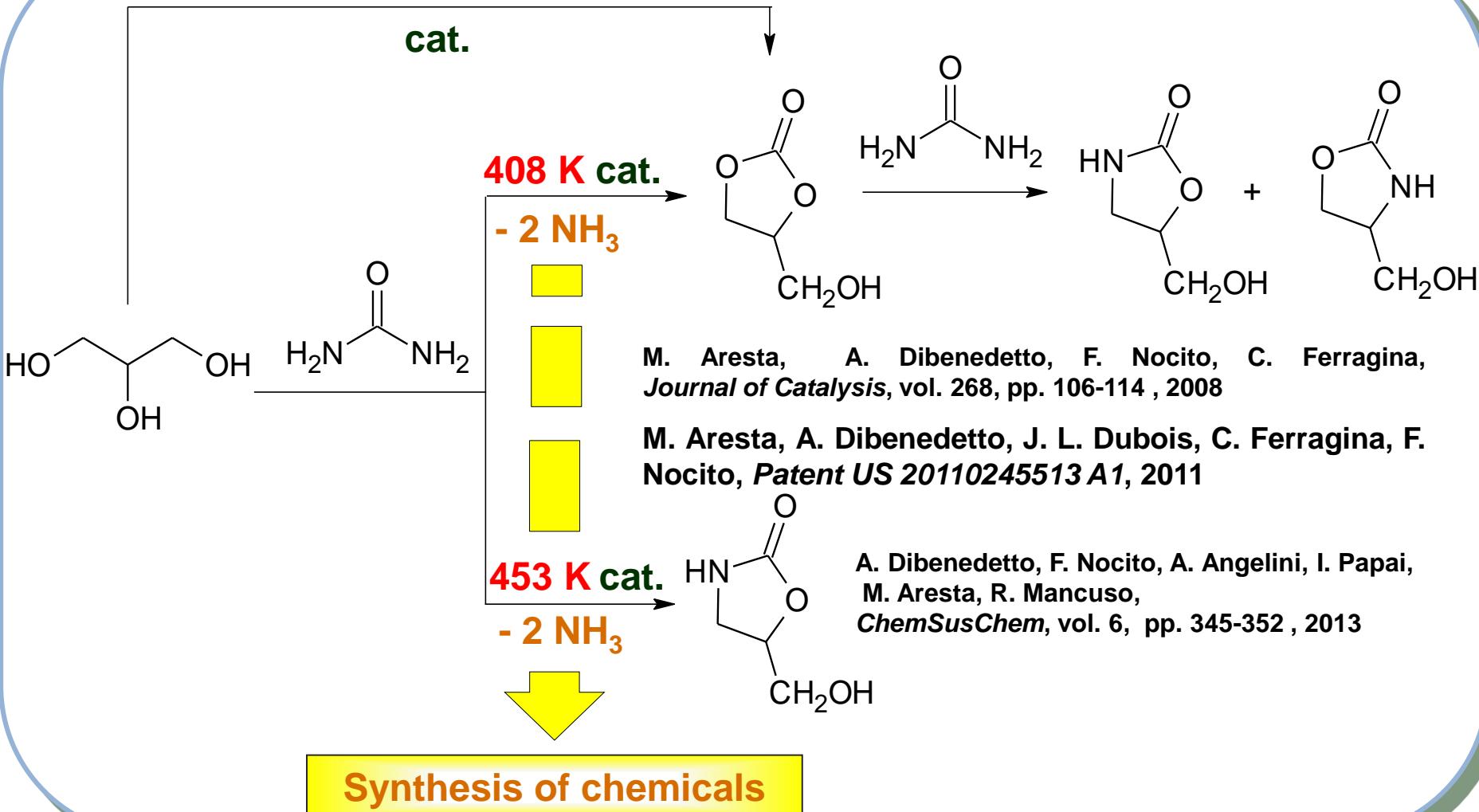
Main industrial applications of glycerol.



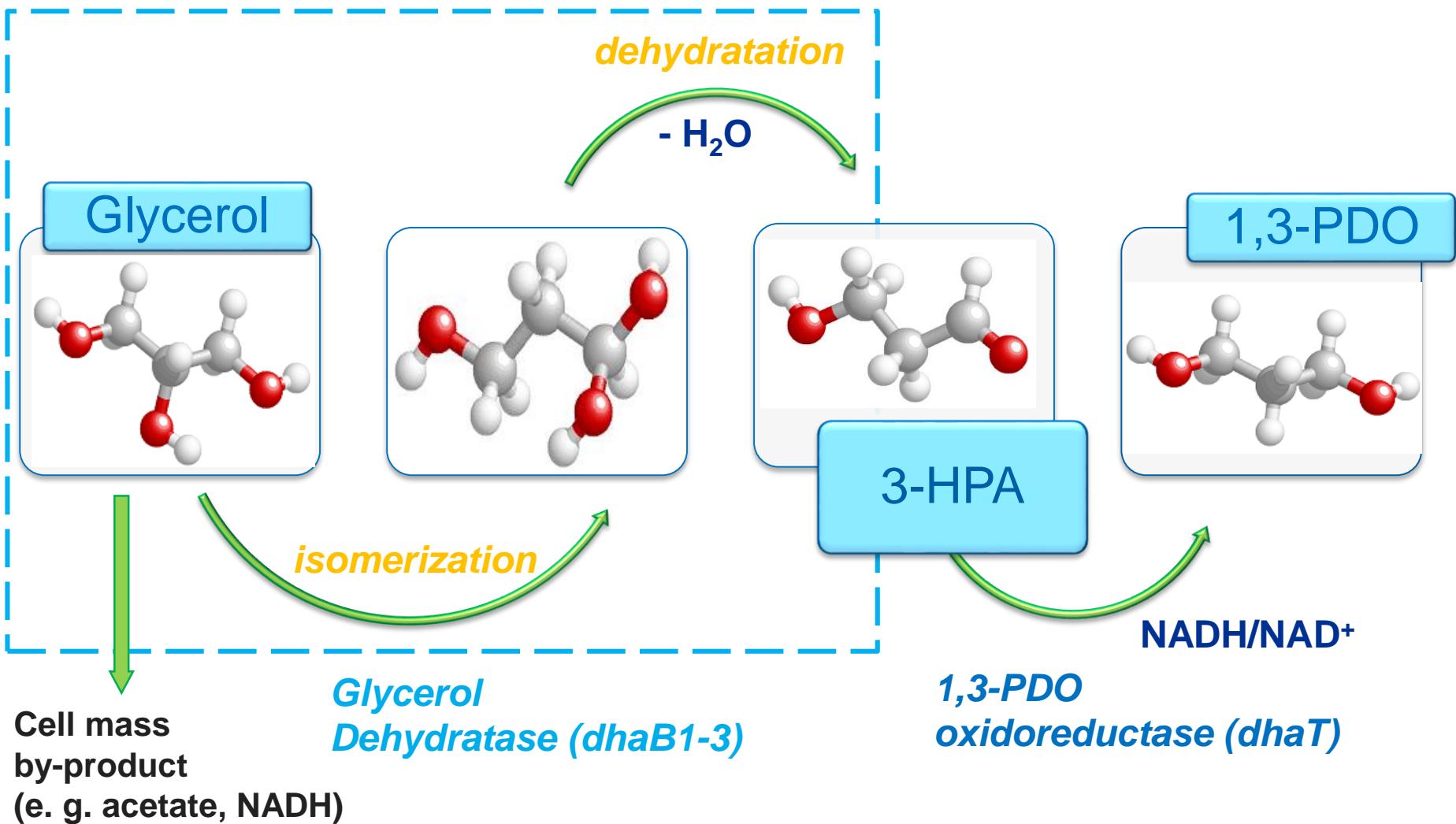
Glycerol conversion



408 K CO₂



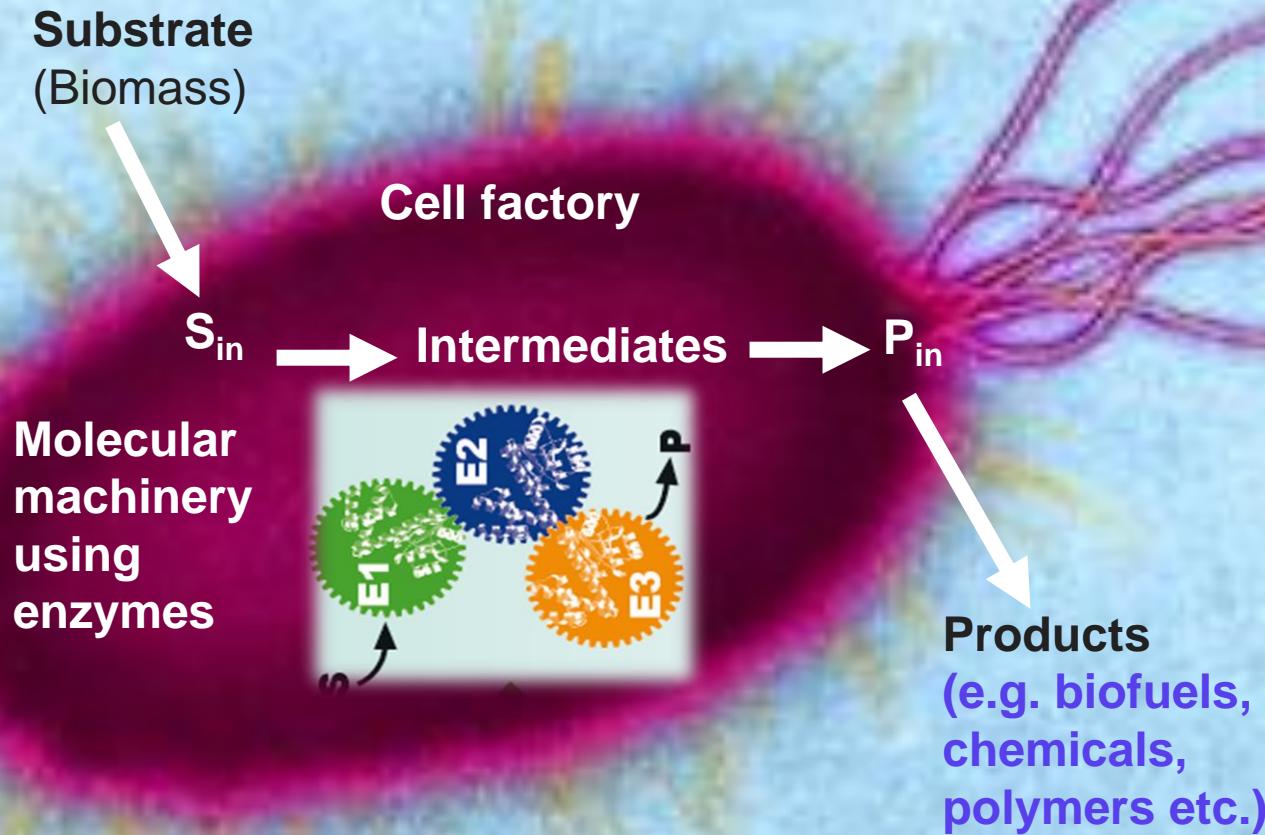
Conversion of glycerol into 1,3-PDO



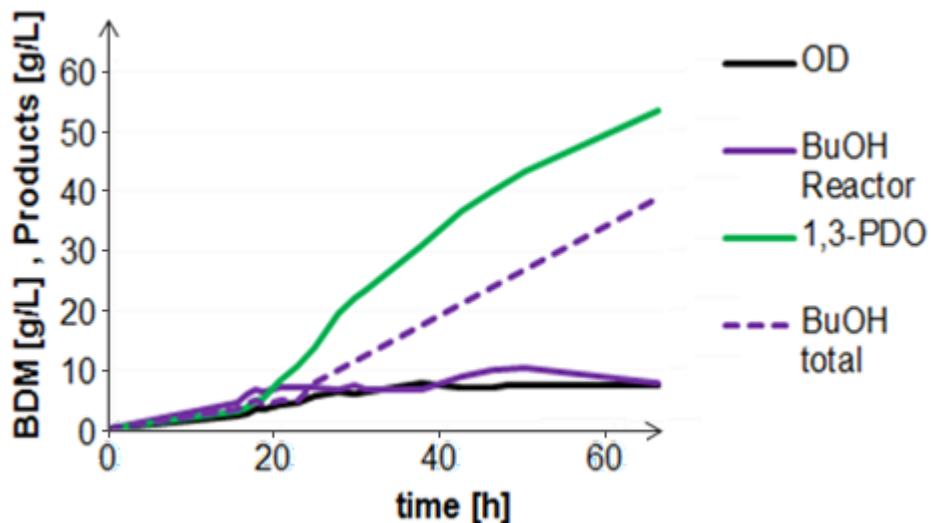
Cell mass
by-product
(e. g. acetate, NADH)

**Glycerol
Dehydratase (*dhaB1-3*)**

**1,3-PDO
oxidoreductase (*dhaT*)**

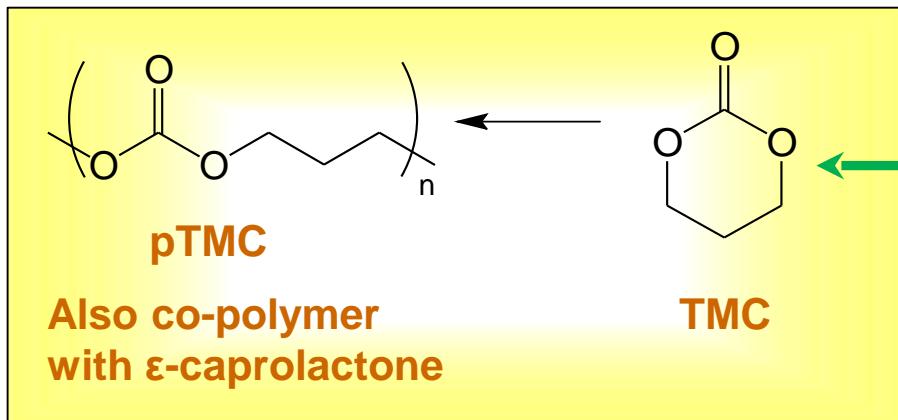


Bioproduction of 1,3-propanediol, n-butanol and biogas from raw glycerol and biomass hydrolysates



An adapted strain for simultaneous use of raw glycerol and hydrolysates. Co-production of **PDO** and **butanol**. In situ removal of butanol to reduce product inhibition





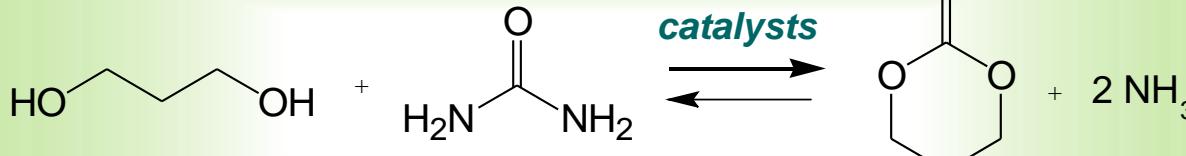
Conversion of 1,3-PDO into TMC

Applications

- Biodegradable materials
- Polymers
- Film and coating materials

Eco-friendly route

New catalysts



Utilization

Benefits

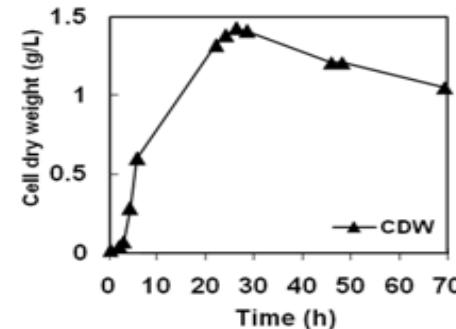
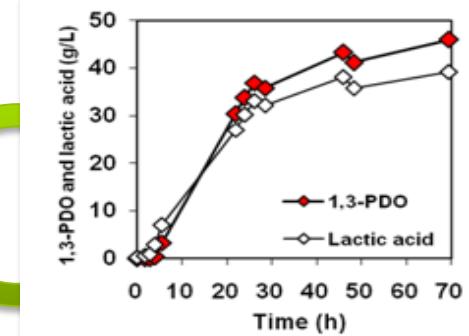
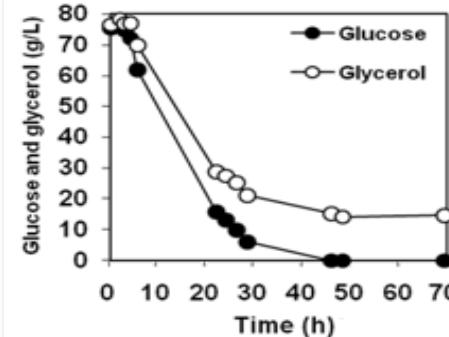
- Absence of reaction solvent
- Low costs of the reagents
- Catalysts recovery and recycle

M. Aresta, A. Dibenedetto, L. di Bitonto, J. L. Dubois, "Synthesis process of Trimethylene carbonate from 1,3-propanediol and urea by heterogeneous catalysts", Patent EP 13192912.7, 2013

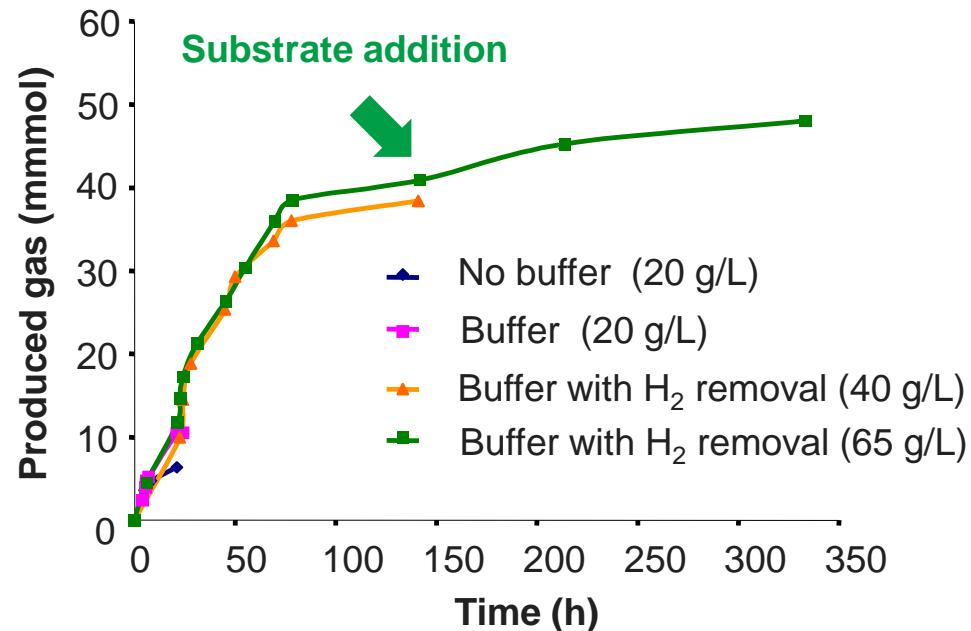
Conversion of glycerol into PDO using *Lactobacillus reuteri* DSM 20016

Glycerol-glucose co-fermentation

- ✓ 46 g/L of PDO obtained (33 g/L using raw glycerol)
- ✓ Yield of 0.9 mol 1,3-PDO/mol glycerol
- ✓ *L. reuteri* DSM 20016 is a valuable probiotic. Biomass can be used.
- ✓ Possible to use raw glycerol and cellulosic hydrolysates as substrates



Hydrogen production from raw bioglycerol using strain ADK1: utilization for the hydrogenation of polyenes



Test conditions (g/L)	% Substrate consumption	% gas produced*	H ₂ (not optimized, mmol/L·h)	P _{max} (atm)
No buffer (20)	88	58	14.28	3.2
Buffer (20)	100	100	25.87	5.3
Buffer with H ₂ removal (40)	74	44	13.4	3.2
Buffer with H ₂ removal (65)	64	34	14.38	3.6

* Respect to initial substrate 1:1 ratio

Chemical (VPR) vs biotechnological H₂ production from bioglycerol

Comparison of the **biological*** vs **catalytic#** production di **dihydrogen** from glycerol

Concentration of glycerol	2-6 %*	1 – 20 %#
Conversion of glycerol	100% at 2% feed	100% at 1% feed
Purity of H ₂	> 99%	90%
Presence of CO	absent	yes
Presence of CO ₂	traces	present
Temperature	ambient	500–600 K
Pressure	0.6 MPa	2.0–3.0 MPa
Lifetime of the catalyst	more than seven days	one week
Co-products	organic acids, ethanol: only traces	organic acids and others

CONCLUSIONS

- **Integration of biotechnology and catalysis may bring to the valorization of waste streams, solid-liquid-gases**
- **The integration can be: “a sequence of operations” for the production/conversion of products, or “integrated technologies” for single product conversion**
- **This zero-waste approach, with recovery/reuse of co-products (ammonia) and use of recoverable and reusable heterogeneous catalysts, converts waste (liquid, solid, gases) into added value products**
- **This concept can find wide implementation**



Acknowledgements

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